

## FEDERAL PUBLIC SERVICE COMMISSION COMPETITIVE EXAMINATION-2020 FOR RECRUITMENT TO POSTS IN BS-17 UNDER THE FEDERAL GOVERNMENT

Roll Number

## PHYSICS, PAPER-I

FHYSICS, FAFER-I								
		LOWED (CQS):	: THREE MAXIM	HOURS UM 30 MIN	UTES	PART-I (MCQS) PART-II	) MAXIMUM MARI MAXIMUM MARI	
<ul> <li>NOTE: (i) Part-II is to be attempted on the separate Answer Book.</li> <li>(ii) Attempt ONLY FOUR questions from PART-II. ALL questions carry EQUAL marks.</li> <li>(iii) All the parts (if any) of each Question must be attempted at one place instead of at different places.</li> <li>(iv) Write Q. No. in the Answer Book in accordance with Q. No. in the Q.Paper.</li> <li>(v) No Page/Space be left blank between the answers. All the blank pages of Answer Book must be crossed.</li> <li>(vi) Extra attempt of any question or any part of the question will not be considered.</li> <li>(vii) Use of Calculator is allowed.</li> </ul>								
<u>PART – II</u>								
Q. 2.	(a) (b)			le product? S	Show that	ain its physical signat $\vec{B} - (\vec{A} \cdot \vec{B})\vec{C}$	nificance.	(10) (6)
	( <b>c</b> )	If $\phi = 2$	$2x^3y^2z^4$ the	n find the div	` '			(4) (20)
Q. 3.	(a) (b)	A space	eship of n		$.50 \times 1$	03 kg is in a circ	cular Earth orbit of radius	(8) (6)
		the forv the peri	ward direct od <i>T</i> of the	ion to decreated resulting ell	ase the siptical of	speed to 96.0% of rbit?	s when a thruster is fired in the original speed. What is	
	(c)	center)	associated	with its rota	tion on	its axis or the angu	of the Earth (relative to its alar momentum of the Earth al motion around the Sun?	(6) <b>(20</b> )
Q. 4.	(a) (b)	Explain	two tests o		on i.e mi	croscopic and mac	-	(6) (8)
	(c)	lifetime measure	e of high-s ed to be 16	peed muons	in a b five sign	urst of cosmic ray	b be 2.2000 ms. The mean ys observed from Earth is nat is the speed parameter $b$	(6) <b>(20</b> )
Q. 5.	(a) (b)						temperature on viscosity? room temperature, is forced	(8) (5)
		through 950 Pa. emergir s, a tota	a pipe of of The pipe of from the al of 1.23	circular cross has a diame free end of	s section eter of 2 the pipe	by a pump that ma .6 cm and a lengt at atmospheric pro-	aintains a gauge pressure of h of 65 cm. The castor oil essure is collected. After 90 efficient of viscosity of the	
	(c)	A liquid bends the horizon	d flows the upward the tal pipe of	rough a hori rough a hei	ght of 6.14 cm	11.5 m where it . What must the vo	adius is 2.52 cm. The pipe widens and joins another plume flux be if the pressure	(7) <b>(20)</b>
Q. 6.	<b>(a)</b>	What is solutior	-	armonic osci	llator? V	Vrite its equation of	f motion and find its	(10)
	<b>(b)</b>	The am What pe	plitude of ercentage of	of the mechar	nical ene	rgy of the oscillato	by 3.0% during each cycle. or is lost in each cycle?	(4)
	(c)	water a very slo	nd hot plat owly to 10	e being initia	ally at 20 h point	0°C. The temperatu the water begins t	ed on a hot plate, both the ure of the hot plate is raised o boil. What is the entropy	(6) <b>(20</b> )

## **PHYSICS, PAPER-I**

- **Q.7.** (a) What are travelling waves? Find the rate at which energy is transported by a wave (5) travelling along a string.
  - (b) A string has linear density  $\mu = 525$  g/m and is under tension T = 45 N. We send a (5) sinusoidal wave with frequency f = 120 Hz and amplitude  $y_m = 8.5$  mm along the string. At what average rate does the wave transport energy?
  - (c) Two sinusoidal waves with the identical wavelengths and amplitudes travel in (10) (20) opposite directions along a string with a speed of 10 cm/s. If the time interval between instants when the string is flat is 0.50 s, what is the wavelength of the waves?
- **Q. 8.** (a) Explain the volume and pressure corrections in ideal gas law as suggested by van (10) der Waals.
  - (b) For oxygen the van der Waals coefficients have been measured to be  $a = 0.138 \text{ J} \cdot \text{m}^3/\text{mol}^2$  and  $b = 3.18 \times 10^{-5} \text{ m}^3/\text{mol}$ . Assume that 1.00 mol of oxygen at T = 50 K is confined to a box of volume 0.0224 m<sup>3</sup>. What pressure does the gas exert according to (a) the ideal gas law and (b) the van der Waals equation?
  - (c) State and explain the zeroth law of thermodynamics.

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(5) (20)

(5)